

High-Throughput Screening

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Direct Surface-Enhanced Raman Scattering Analysis of DNA **Duplexes****

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Abstract: The exploration of the genetic information carried by DNA has become a major scientific challenge. Routine DNA analysis, such as PCR, still suffers from important intrinsic limitations. Surface-enhanced Raman spectroscopy (SERS) has emerged as an outstanding opportunity for the development of DNA analysis, but its application to duplexes (dsDNA) has been largely hampered by reproducibility and/or sensitivity issues. A simple strategy is presented to perform ultrasensitive direct label-free analysis of unmodified dsDNA with the means of SERS by using positively charged silver colloids. Electrostatic adhesion of DNA promotes nanoparticle aggregation into stable clusters yielding intense and reproducible SERS spectra at nanogram level. As potential applications, we report the quantitative recognition of hybridization events as well as the first examples of SERS recognition of single base mismatches and base methylations (5-methylated cytosine and N6-methylated Adenine) in duplexes.

DNA is the custodian of the memory of life. Although very stable, it is subject to modifications that are associated with evolution but more frequently with genetic diseases or cancer.[1] Owing to the small content of DNA in cells and its structural complexity, its analysis require denaturation (heat), fragmentation (restriction endonucleases), separation (electrophoresis), amplification (polymerase chain reaction, PCR) and detection (quantitative PCR or microarray techniques).^[2] Besides the cost both in time and money, during these processes samples are subject to alterations^[3] and losses of epigenetic information.^[4] The advent of nanophotonics offers a unique opportunity for the development of direct, fast, and ultrasensitive methods for DNA analysis.^[5] In the broad field of plasmonics, surface-enhanced Raman scattering (SERS) spectroscopy has arisen as a powerful analytical tool for detection and structural characterization of biomolecules.^[6] The large majority of SERS-based DNA detection strategies rely on the selective recognition between two complementary strands to identify a specific sequence, and the use of extrinsic Raman labels for SERS readout.^[7] However, labeling of the DNA strands requires complex and costly chemistry, and does not provide any chemical-specific information.^[8] A second approach is based on the direct detection of the distinctive SERS signal from DNA strands directly adsorbed onto the nanostructured surface. This strategy shows outstanding potential in terms of sensitivity, selectivity, and chemicalspecific information.^[8,9] To date, however, direct SERS analysis of unmodified DNA is mainly restricted to singlestranded DNA sequences (ssDNA), as the direct contact of double-stranded sequences (dsDNA) with nanostructured metallic surfaces (negative at the physiological pH) is largely hindered by the negative charge of the phosphate backbone, unless relatively high DNA concentrations are employed. [9b,c] As a result, dsDNA SERS spectra traditionally suffer from inherent poor spectral reproducibility^[8,9b] and/or limited sensitivity, hampering the extended application of label-free SERS strategies for the study of unmodified duplexes. Thus, successful translation of the analytical potential of SERS to the direct study of dsDNA would be a great leap forward for the full exploration of the genetic information.

To overcome this major challenge, in this study we use positively charged silver nanoparticles coated with spermine molecules (AgNP@Sp). Spermine bound to AgNPs acts as stabilizer, and upon addition of negatively-charged DNA, it promotes NP aggregation into stable particle clusters.[10] This has important consequences for the direct SERS analysis of

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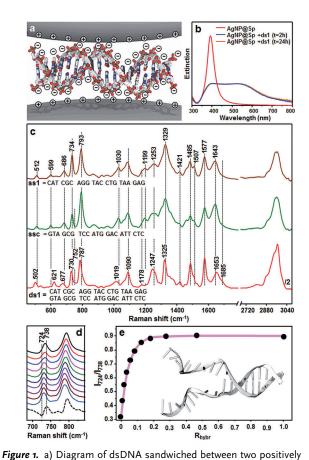
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Supporting information for this article (experimental, AgNP@Sp. characterization (Section S1), DNA-driven aggregation of AgNP@Sp and spectral reproducibility (Section S2), vibrational assignment of dsDNA SERS spectra (Section S3), SERS of ss and ds sequences on silver hydroxylamine colloids (Section S4), ds1 SERS spectrum and the pondered sum of the individual spectra of ss1 and ssc (Section S5), and SERS spectra of ds2, ds3 and ds4 (Section S6)) is available on the WWW under http://dx.doi.org/10.1002/ anie.201408558.

dsDNA. Firstly, the DNA adsorption occurs by non-specific electrostatic interaction of the phosphate groups rather than by base-specific nucleoside-metal binding. Secondly, DNA promotes the nanoparticle aggregation removing the need of external aggregating agents. On the one hand, this significantly simplifies the sensing scheme, reducing the number of variables to be controlled (type and concentration of the aggregating agent, time and order of addition, and so on). On the other hand, DNA sequences are selectively sandwiched between interparticle junctions (that is, hot-spots^[11]), thus maximizing the intensification of the Raman signal (Figure 1a). Thirdly, the nanoparticle aggregation occurs fast and independently of the base composition, generating long-term stable clusters in suspension, as demonstrated by the extinc-



charged AgNP@Sp. b) Extinction spectra of pure AgNP@Sp colloids and in the presence of ds1 (final conc. 630 ng mL⁻¹) acquired 2 h and 24 h upon DNA addition (the AgNP@Sp+ds1 mixture was sonicated for few seconds before the spectral acquisition). c) SERS spectra of ss1 and ssc (3.15 μ g mL⁻¹), and ds1 (630 ng mL⁻¹) on AgNP@Sp. SERS spectra were normalized to the PO₂⁻ stretching at 1090 cm⁻¹. The ds1 SERS spectrum in the 2700-3050 cm⁻¹ region was divided by two. d) Detail of the 680-890 cm⁻¹ spectral region for the SERS spectra of ssc+ds1 mixtures at different molar ratios $R_{hybr} = [ds1]/([ds1] + [ssc])$ (from top to bottom: 0.011, 0.024, 0.041, 0.063, 0.091, 0.130, 0.189, 0.286, 0.474, and 1). The ds1 concentration was progressively increased from 0 to 630 $\mbox{ng}\,\mbox{mL}^{-1}$ while the ssc concentration was simultaneously decreased from 3.15 $\mu g\,mL^{-1}\,ng$ to zero. The dotted line is the difference spectrum ssc-ds1 obtained by digitally subtracting the SERS spectra shown in (c). e) Ratiometric peak intensities I_{724}/I_{738} vs. $R_{hybr} = [ds1]/([ds1] + [ssc])$ molar ratio.

tion spectra illustrated in Figure 1b. This also allows us to perform the SERS measurements after several hours upon the analyte addition (that is, when the process of DNA adhesion has reached its equilibrium). Finally, SERS spectra are acquired in colloidal suspensions with a long working distance objective. In this experimental set-up, the recorded SERS spectra are the statistically-averaged result of a large ensemble of nanoparticle/DNA clusters in continuous Brownian motions within the scattering volume. Such averaged bulk SERS regime yields good-quality spectra with welldefined average band centers, bandwidths, and relative intensities.[12] As a result, dsDNA electrostatic adhesion onto the AgNP@Sp nanoparticles allows for the ultrasensitive analysis of duplexes and their modifications at the nanogram regime with high batch-to-batch reproducibility, yielding SERS spectra that are truly spectral representations of the nucleoside composition and organization within the helix. To confirm this, we show an ample set of examples of direct labelfree SERS detection of unmodified duplexes, including quantification of hybridization events as well as recognition of single-base mismatches and base methylations (5-methylated cytosine and N6-methylated adenine), which to date were restricted to single-stranded sequences. [9a,d]

The average SERS spectra of two complementary singlestranded sequences (ss1 and ssc) and their corresponding double-helix structure (ds1) on AgNP@Sp colloids are shown in Figure 1c. The overall dsDNA concentration was kept constant throughout the entire study at ca. 630 ng mL-(corresponding to less than 3.0 ng in the probed volume; see the Supporting Information, Section S2) and appropriately selected to yield long-term stable clusters in suspension with reproducible and unvaried SERS spectral profiles (Section S2). Vibrational assignment of dsDNA was based on literature references and comparative spectral analysis with homo- and bi-polymeric sequences (Section S3). In contrast with previous results on negatively charged colloids (Section S4), [13] when the individual ss units hybridize into the ds a large reshaping of their SERS spectra is observed (Figure 1c), influencing peak position, bandwidths, and relative intensity. Notably, if ss1 and ssc spectra show small changes in relative intensity that are associated with the base composition, the stacking and pairing of the nucleobases into the hybridized native structure is revealed by several characteristic features of the Raman melting profiles.^[14] Among others, we highlight the marked intensity decrease and peak-shifting of the ring breathing modes (700-800 cm⁻¹) and the carbonyl stretching modes (1653 cm⁻¹), the latter being extremely sensitive to the disruption of Watson-Crick hydrogen bonding.[14] We also observe a general narrowing of the Raman features in the 1150-1600 cm⁻¹ spectral region, containing many overlapping bands mainly arising from in-plane vibrations of base residues.^[15] Such spectral modification can be ascribed to the adoption of a more rigid and well-defined conformation by the two individual ss when they self-organize into the rigid duplex geometry. The investigation of a set of solutions containing different ds1 and ssc molar ratios, R_{hvbr} = [ds1]/([ds1] + [ssc]) (Figure 1d) shows a progressive shift of the ring breathing bands of adenine (from 734 cm⁻¹ to 730 cm⁻¹) when the DNA population is enriched with ds1.

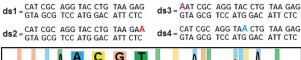
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The spectral subtraction of the SERS spectra allows us to fully disclose the spectral alterations associated with changes in the DNA structure. In particular, the difference spectrum ds1-ssc (dotted line, Figure 1 d) reveals a minimum at 724 cm⁻¹ and a maximum at 738 cm⁻¹. The corresponding ratiometric peak intensities I_{724}/I_{738} were then selected to monitor the relative ds1/ssc populations in the samples, and plotted against $R_{\rm hybr}$ (Figure 1 e). Data shows that ds sequences can be clearly identified within the mixture even when present at values <1%, whereas SERS spectra are fully dominated by ds1 contributions for duplex populations \geq 20%. Such result is consistent with the higher affinity of dsDNA to AgNP@Sp owing to the larger availability of negatively charged phosphate groups as compared to ssDNA. Linear correlation ($r^2 > 0.99$) is observed in the $0.01-0.19\,R_{\rm hybr}$ range.

To test the possibility of detecting single-base mismatches in DNA duplexes, SERS spectra of the full-complementary ds1 and the heteroduplexes ds2, ds3, and ds4 were acquired and compared (Figure 2). These heteroduplexes contain one

Single-base mismatch



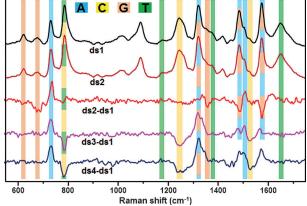


Figure 2. SERS spectra of ds1 and ds2; and digitally subtracted SERS spectra ds2-ds1, ds3-ds1, and ds4-ds1. For the sake of clarity, the digitally subtracted spectra were multiplied by 3 (for ds2-ds1) and 4 (for ds3-ds1 and ds4-ds1). Frequency positions of characteristic nucleotide bands are highlighted in: blue (adenine), yellow (cytosine), orange (guanine), and green (thymine). The overall ds concentration was kept constant at 630 ng mL $^{-1}$.

adenine base, A, in place of: (ds2) one guanine, G, (ds3) one cytosine, C, (terminal position), and (ds4) one cytosine (internal position). Subtraction of the SERS spectra of ds1 from the other samples generates difference spectra containing vibrational signatures associated with the additional (positive features) and removed nucleobase (negative features). Positive A bands emerge in all difference spectra at 730 and 1507 cm⁻¹, whereas negative G features are observed at ca. 620, 675, and 1354 cm⁻¹ in the ds2—ds1. In this case, negative bands also appear at 1487 and 1577 cm⁻¹, ascribed to purine modes (mainly G contribution) while the other purine feature at 1325 cm⁻¹ (mainly A contribution) does not suffer

from a marked change in the relative intensity. These results, together with the lack of significant changes in SERS intensity of the pyrimidine bases, clearly indicate that the spectral alterations are ascribed to the A→G base-mismatch in the ds2. On the other hand, ds3-ds1 and ds4-ds1 difference spectra show very similar spectral patterns where, in addition to the positive A (730 and 1507 cm⁻¹) and the negative C (1250 and 1528 cm⁻¹) contributions, a consistent intensity increase of the purine bands (1325, 1487, and 1577 cm⁻¹) is observed. The pyrimidine ring breathing $(787 \text{ cm}^{-1}, \text{ C} + \text{T})$ also undergoes a drastic intensity decrease whereas thymine marker bands do not reveal significant alterations. These spectral changes can therefore be associated with the $A \rightarrow C$ base-mismatch in ds3 and ds4. Minor differences between ds3 and ds4 difference spectra (Figure 2) can be ascribed to the different position of the base mismatch within the sequence.[15]

The potential application of AgNP@Sp in SERS analysis of 5-methylated cytosine bases (mC) and N6-methylated adenine bases (mA) within ds structures was examined by acquiring the SERS spectra of dsmC and dsmA and comparing to their non-methylated analogous ds1. In dsmC and dsmA, the C or A nucleobases of one of the strands were all replaced by their respective 5-methylated and N6-methylated counterparts (see sequence structures in Figure 3 and Figure 4). All

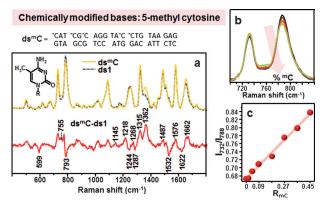


Figure 3. a) Molecular structure of 5-methyl cytosine ($^{\text{m}}$ C). SERS spectra of ds1 and ds $^{\text{m}}$ C, and the difference spectrum ds $^{\text{m}}$ C-ds1. For the sake of clarity, the digitally subtracted spectrum was multiplied by a factor of two. b) Detail of the 700–840 cm $^{-1}$ region for the SERS spectra of ds1 + ds $^{\text{m}}$ C at different cytosine base ratios $R_{\text{mc}} = [^{\text{m}}\text{C}]/([C] + [^{\text{m}}\text{C}])$ (from the top to the bottom, $R_{\text{mc}} = 0$, 0.045, 0.091, 0.182, 0.273, 0.364, and 0.455). The overall ds concentration was kept constant at 630 ng mL $^{-1}$. c) Ratiometric peak intensities I_{732}/I_{788} vs. the molar ratio $R_{\text{mc}} = [^{\text{m}}\text{C}]/([C] + [^{\text{m}}\text{C}])$.

the spectral changes are in good agreement with the normal Raman studies of the substitution of the methyl group into the 5C and N6 atom in cytosine^[16] and adenosine^[17] nucleosides, respectively. The major spectral markers of the cytosine methylation can be easily recognized through the whole difference spectrum (Figure 3a). Among others, we observe a red-shift and relative intensity decrease of the pyrimidine ring breathing band at 787 cm⁻¹, together with the appearance of a new weak feature at ca. 755 cm⁻¹. In the 1200–1450 cm⁻¹

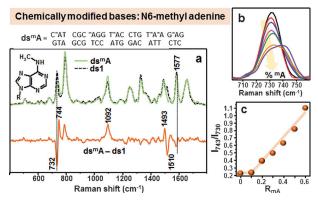


Figure 4. a) Molecular structure of N6-methyladenine (^{m}A). SERS spectra of ds1 and ds ^{m}A , and the difference spectrum ds ^{m}A —ds1. For the sake of clarity, the digitally subtracted spectrum was multiplied by two. b) Detail of the 700–760 cm $^{-1}$ region for the SERS spectra of ds1 + ds ^{m}A at different adenine base ratios $R_{mA} = [^{m}A]/([A] + [^{m}A])$ (from the top to the bottom $R_{mA} = 0$, 0.1, 0.2, 0.3, 0.4, 0.5, and 0.6). The overall ds concentration was kept constant at 630 ng mL $^{-1}$. c) Ratiometric peak intensities I_{743}/I_{730} vs. molar ratio R_{mA} .

spectral region, the normal Raman of unmodified cytidine aqueous solution contains two main bands at 1242 and 1292 cm⁻¹, whereas the 5-methyl substituted shows a very different spectral signature with three discernible bands at 1224, 1258, and 1302 cm⁻¹ and a new very intense feature at 1362 cm⁻¹. [16] Accordingly, the difference spectrum in Figure 3 a reveals two negative bands at ca. 1244 and 1287 cm⁻¹; three positive contributions broadly centered at 1218, 1268, and 1315 cm⁻¹, and a very strong and sharp band at 1362 cm⁻¹. Moreover, the C ring band at about 1510 cm⁻¹ also undergoes a marked red-shift and intensity decrease while the broad carbonyl stretching band centered at 1653 cm⁻¹ shows a large red-shift (well highlighted in the difference spectrum by the negative feature at ca. 1622 cm⁻¹ and the positive one at ca. 1662 cm⁻¹).

The spectral modifications of the SERS spectral profile of the normal DNA upon methylation of six of their eleven the A bases are also striking (Figure 4a, difference spectrum). The A ring breathing band at 731 cm $^{-1}$ undergoes a drastic 11 nm up-shift together with a notable intensity decrease, whereas the pyrimidine stretching of the adenine at 1509 cm $^{-1}$ almost disappears from the spectrum. Furthermore, we observe up-shifts of the phosphate stretching mode at 1090 cm $^{-1}$ (together with a marked intensity increase); the v(CN) imidazole feature at 1487 cm $^{-1}$ and the 1577 cm $^{-1}$ band ascribed to base ring modes (mainly G+A).

Quantification of relative methylated base populations within the sample was investigated by monitoring the ratiometric peak intensities I_{732}/I_{788} for $^{\rm m}{\rm C}$ and I_{743}/I_{730} for $^{\rm m}{\rm A}$, at different molar ratios ($R_{\rm mC} = [^{\rm m}{\rm C}]/([{\rm C}] + [^{\rm m}{\rm C}])$) and $R_{\rm mA} = [^{\rm m}{\rm A}]/([{\rm A}] + [^{\rm m}{\rm A}]))$. Details of the ring breathing spectral regions are illustrated in Figure 3b and Figure 4b. In the case of ds $^{\rm m}{\rm C}$ (Figure 3b), we highlight the progressive intensity decrease of the C+T ring breathing band at 787 cm $^{-1}$ upon increasing of the relative $^{\rm m}{\rm C}$ populations. Differently, for ds $^{\rm m}{\rm A}$ (Figure 4b), the A ring breathing band at 730 cm $^{-1}$ suffers a dramatic intensity decrease and large

peak red-shift as the $R_{\rm mA}$ value becomes larger. Linear correlations were obtained in both cases ($r^2 > 0.99$ for ^mA and > 0.94 for ^mC) with detection limits of one ^mC per 22 total cytosine bases and one ^mA per 10 total adenine bases in the sample (Figure 3c and Figure 4c, respectively).

In summary, we have demonstrated the potential application of SERS for the development of a fast and affordable high-throughput screening method for gaining detailed genomic information on DNA duplexes. Importantly, owing to the low amount of DNA required (comparable to only 10–100 cells), the analysis can be performed without the necessity of amplification, thus providing realistic direct information of the DNA in its native state. Owing to the direct nature of this sensitive and selective assay, we anticipate these findings of key importance in a number of different fields including diagnosis, genetic engineering, biotechnology, drug discovery (antibiogram in microbiology), agriculture, and forensic science.^[1,18]

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- [1] a) L. Pikor, K. Thu, E. Vucic, W. Lam, Cancer Metastasis Rev. 2013, 32, 341–352; b) S. P. Jackson, J. Bartek, Nature 2009, 461, 1071–1078.
- [2] a) S. Yang, R. E. Rothman, Lancet Infect. Dis. 2004, 4, 337–348;
 b) A. Git, H. Dvinge, M. Salmon-Divon, M. Osborne, C. Kutter, J. Hadfield, P. Bertone, C. Caldas, RNA-Publ. RNA Soc. 2010, 16, 991–1006.
- [3] X. Qiu, L. Wu, H. Huang, P. E. McDonel, A. V. Palumbo, J. M. Tiedje, J. Zhou, *Appl. Environ. Microbiol.* 2001, 67, 880–887.
- [4] E. E. Schadt, S. Turner, A. Kasarskis, *Human Molec. Genetics* 2010, 19, R227 – R240.
- [5] a) J. N. Anker, W. P. Hall, O. Lyandres, N. C. Shah, J. Zhao, R. P. Van Duyne, *Nat. Mater.* **2008**, *7*, 442–453; b) K. Saha, S. S. Agasti, C. Kim, X. N. Li, V. M. Rotello, *Chem. Rev.* **2012**, *112*, 2739–2779.
- [6] a) R. A. Alvarez-Puebla, L. M. Liz-Marzan, Small 2010, 6, 604 610; b) W. Xie, S. Schlücker, Phys. Chem. Chem. Phys. 2013, 15, 5329 5344.
- [7] D. Graham, K. Faulds, Chem. Soc. Rev. 2008, 37, 1042-1051.
- [8] A. Barhoumi, D. Zhang, F. Tam, N. J. Halas, J. Am. Chem. Soc. 2008, 130, 5523 – 5529.
- [9] a) A. Barhoumi, N. J. Halas, J. Phys. Chem. Lett. 2011, 2, 3118–3123; b) S. R. Panikkanvalappil, M. A. Mackey, M. A. El-Sayed, J. Am. Chem. Soc. 2013, 135, 4815–4821; c) S. R. Panikkanvalappil, M. A. Mahmoud, M. A. Mackey, M. A. El-Sayed, ACS Nano 2013, 7, 7524–7533; d) E. Papadopoulou, S. E. J. Bell, Angew. Chem. Int. Ed. 2011, 50, 9058–9061; Angew. Chem. 2011, 123, 9224–9227; e) Ž. Krpetić, I. Singh, W. Su, L. Guerrini, K. Faulds, G. A. Burley, D. Graham, J. Am. Chem. Soc. 2012, 134, 8356–8359.
- [10] D. van Lierop, Ž. Krpetić, L. Guerrini, I. A. Larmour, J. A. Dougan, K. Faulds, D. Graham, Chem. Commun. 2012, 48, 8192–8194.
- [11] F. J. García de Abajo, Rev. Mod. Phys. 2007, 79, 1267-1290.
- [12] R. Aroca, Surface-enhanced Vibrational Spectroscopy, Wiley, Chichester. 2006.



- [13] a) E. Papadopoulou, S. E. J. Bell, *Chem. Eur. J.* **2012**, *18*, 5394–5400; b) N. E. Marotta, K. R. Beavers, L. A. Bottomley, *Anal. Chem.* **2013**, *85*, 1440–1446.
- [14] J. G. Duguid, V. A. Bloomfield, J. M. Benevides, G. J. Thomas, *Biophys. J.* 1996, 71, 3350 – 3360.
- [15] H. Deng, V. A. Bloomfield, J. M. Benevides, G. J. Thomas, *Biopolymers* 1999, 50, 656–666.
- [16] A. Gfrörer, M. E. Schnetter, J. Wolfrum, K. O. Greulich, Ber. Bunsen-Ges. 1991, 95, 824–833.
- [17] S. Mansy, W. L. Peticolas, R. S. Tobias, Spectrochim. Acta Part A 1979, 35, 315 – 329.
- [18] a) G. H. Clever, M. Shionoya, Coord. Chem. Rev. 2010, 254, 2391–2402; b) C. J. Lord, A. Ashworth, Nature 2012, 481, 287– 294.